

Hard and Soft Condensed Matter

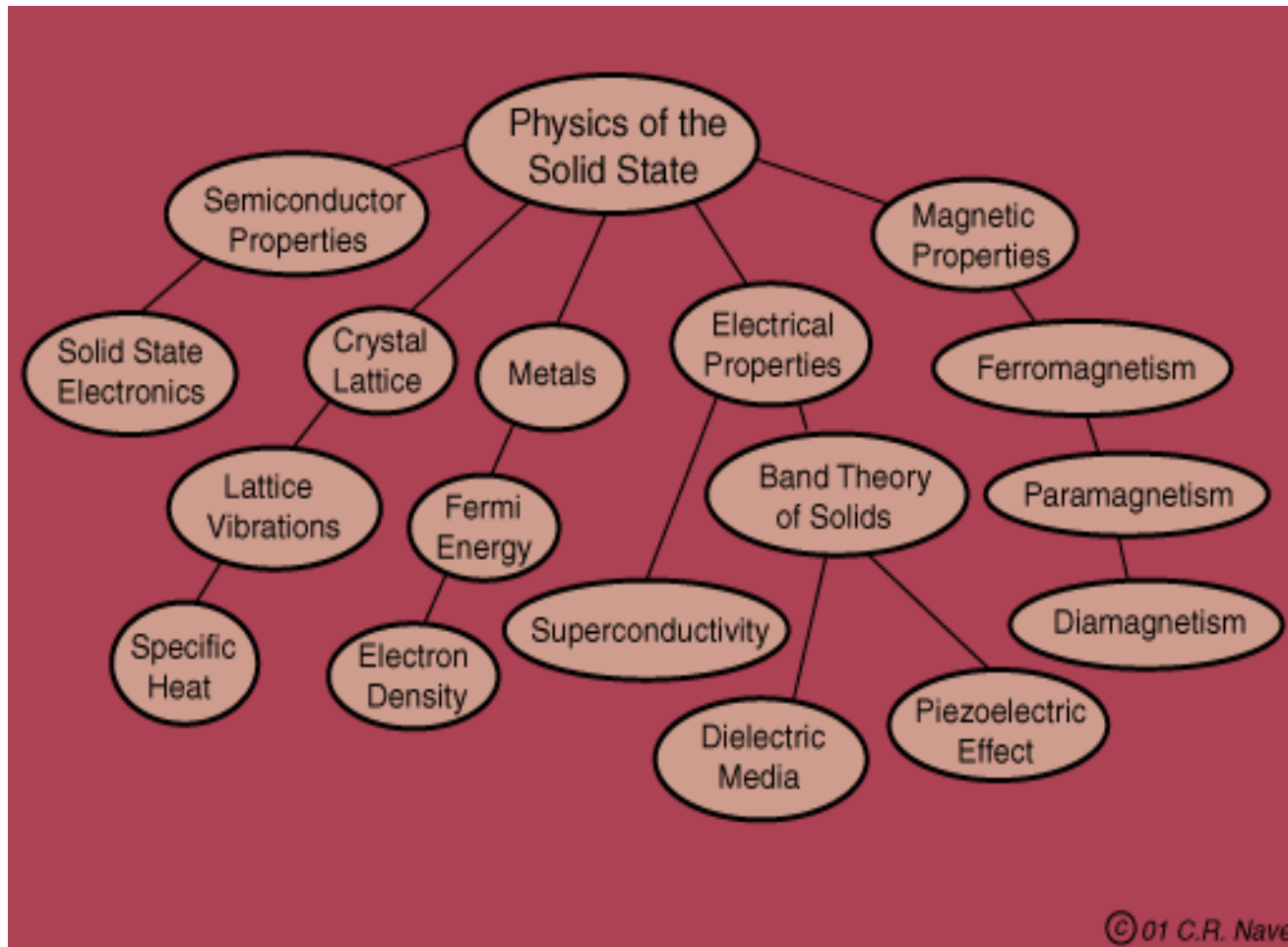


"Hard"
Crystal structure
Heat capacity
Magnetism
Conductivity etc..



"Soft"
Liquid Crystals
Gels, Colloids
Biomaterials etc..

Traditional Topics in Condensed Matter physics



Experimental techniques

- **Scattering:** x-rays, neutrons, electrons
- **Resonance methods:** nuclear magnetic resonance, muon spin resonance
- **Thermodynamic properties:** specific heat, thermal conductivity
- **Electronic properties:** resistivity, Hall effect, photoemission
- **Optical measurements :** microwave, infrared, etc.



Heat capacity (low temperatures)



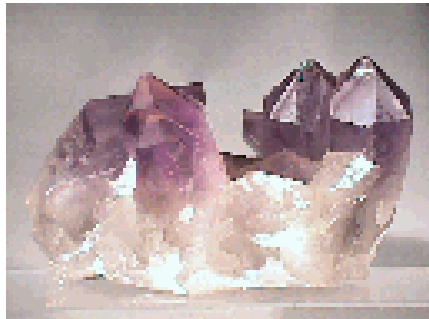
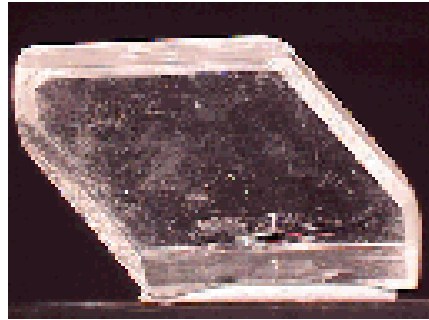
Neutron scattering
(at NIST, USA)

Crystals

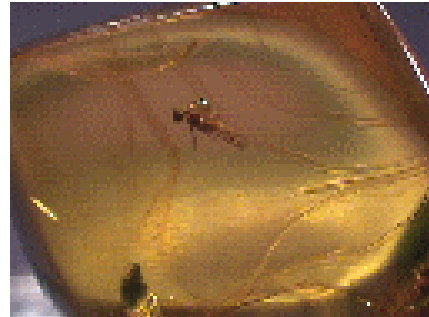
Hexagonal beryl



Monoclinic gypsum



Triagonal quartz



Amorphous amber
(no underlying
crystal symmetry)

Liquid Crystals

The Liquid Crystalline state of matter is somewhere in-between solid crystals and isotropic liquids.

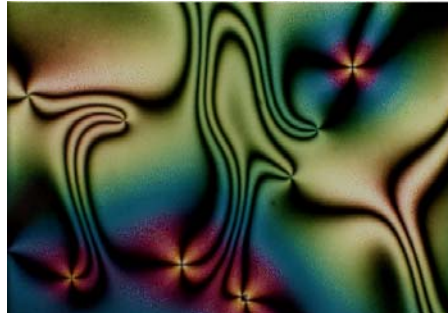
- Molecules have orientational order but do not sit on a lattice
- they can flow like a fluid
 - they are **anisotropic** liquids

Crystal

Liquid Crystal

Liquid

Temp →

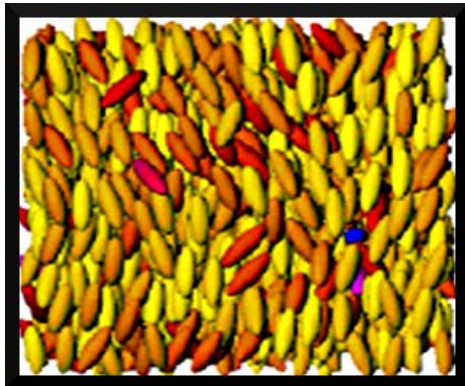


Liquid crystal phases

There are many different liquid crystalline phases, all defined by

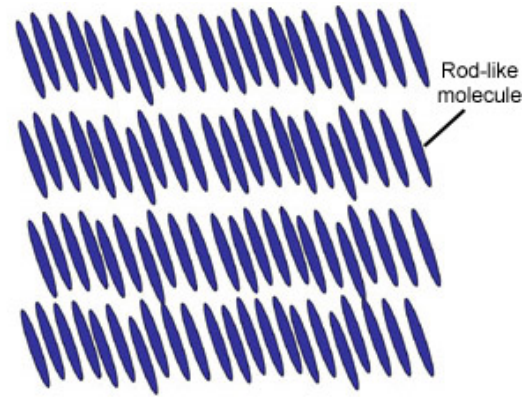
- a) Fluid-like properties
- b) Molecular anisotropy
- c) They are birefringent

The Nematic Phase



A liquid in which all the molecules point in roughly the same direction

The smectic C phase



Liquid-like sheets of molecules arranged in layers